

CHROM. 11,690

Note

Separation of vitamin E isomers by high-performance liquid chromatography

C. C. TANGNEY, J. A. DRISKELL and H. M. McNAIR

Departments of Human Nutrition and Foods and Chemistry, Virginia Polytechnic Institute and State University, Blacksburg, Va. 24061 (U.S.A.)

(First received September 13th, 1978; revised manuscript received December 28th, 1978)

Previous analyses for vitamin E in biological materials have predominately relied upon spectrophotometric^{1,2}, indirectly measuring total tocopherols, thin-layer chromatographic-gas-liquid chromatographic^{3,4} or thin-layer chromatographic-spectrophotometric⁵ combination techniques that are too time-consuming for routine work. Several researchers have reported high-performance liquid chromatographic analyses of mixtures of fat-soluble vitamins^{6,7} but the isolation of the different tocopherols has received little attention. Since it may be important to differentiate between the various tocopherols, particularly in foods, in order to assess vitamin E activity accurately, we propose this simple rapid liquid chromatographic technique for analyses of vitamin E — α -, β -, γ -, and δ -tocopherols.

EXPERIMENTAL

Materials

Hexanes (HPLC grade) and 2-propanol (spectro-grade) were obtained from Fisher Scientific (Pittsburgh, Pa., U.S.A.), α -, γ -, and δ -tocopherols, from Hoffmann-La Roche (Nutley, N.J., U.S.A.), and a β - and γ -tocopherol mixture, from Eisai (Tokyo, Japan).

Apparatus

Spectra-Physics (Santa Clara, Calif., U.S.A.) Models 3500 and 3100 liquid chromatographs equipped with a Model 230 mixed wavelength UV spectrophotometer were employed. A Chromatronix HP sample valve was purchased with the above instrument; the 20- μ l loop was used for initial separations and under calibration conditions, a large loop (30 μ l) was prepared by the investigators. A stainless-steel column (25 cm \times 2.0 mm I.D.) was purchased from Spectra-Physics packed with 5- μ m Spherisorb. The mobile phase was hexane-isopropanol (99.75:0.25) at a flow-rate of 0.8 ml/min. The absorbance at 280 nm was monitored at a chart speed of 1 in./min on a 10-mV linear recorder (Linear Instruments, Irvine, Calif., U.S.A.). Peak heights were measured.

RESULTS AND DISCUSSION

Six standard solutions of each of the four tocopherols were prepared in hexane and analyzed by high-performance liquid chromatography. Solutions were blown down under nitrogen gas immediately following injections and kept in amber-colored sample bottles; when not in use, solutions were stored under nitrogen at 5°. These calibration solutions produced linear calibration curves between 0.03 and 15 μg (Fig. 1).

Good separation of α -, β -, γ - and δ -tocopherols was obtained (Fig. 2). The β - and γ -tocopherol peaks exhibited a resolution of 1.5 with the column yielding 1987

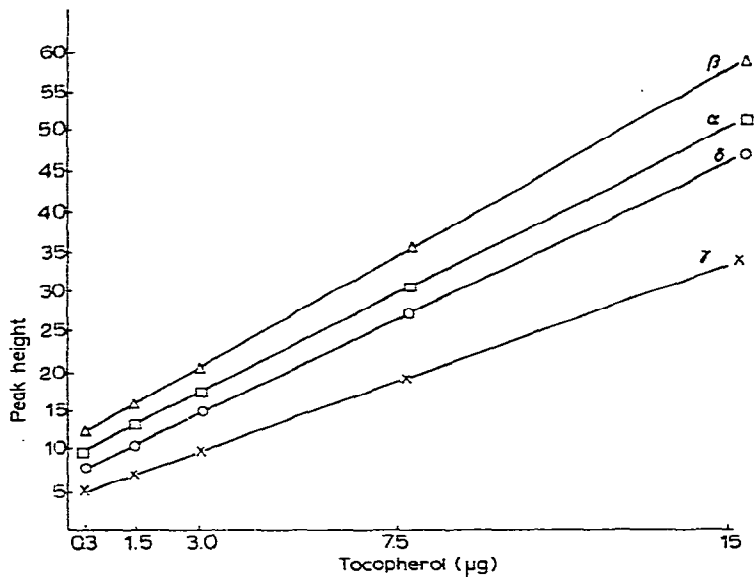


Fig. 1. Ultraviolet detector response for various tocopherols at 280 nm. \square = α -; \triangle = β -; \times = γ -; and \circ = δ - tocopherols.

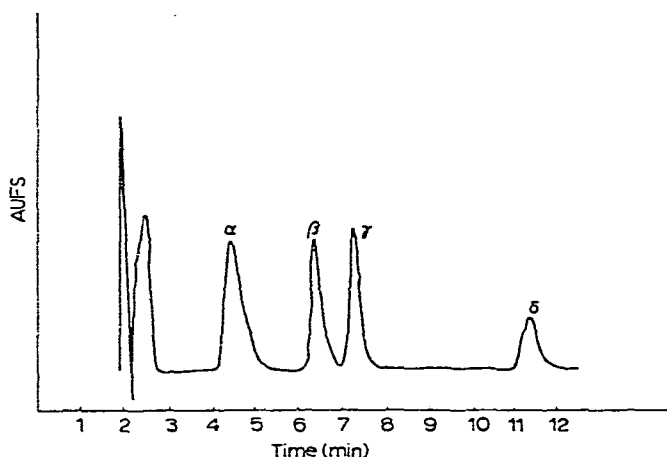


Fig. 2. High-performance liquid chromatogram of tocopherol standards at 280 nm. Experimental conditions are given in the text.

effective plates for the β -tocopherol peak. Retention ratios relative to α -tocopherol were as follows: β -, 1.44; γ -, 1.59; and δ -, 246. The retention time for α -tocopherol was 4.5 min.

The lower limit of detection was found to be approximately 1 μg tocopherol/ml. It has been proposed that smaller quantities of tocopherols would be detected through use of a spectrofluorometer^{8,9}; however with the universal equipment used in this study, one can expect to detect and quantitate the vitamin E compounds in plasma, platelet and food samples, injecting between 10 and 30 μl . With such unsophisticated inexpensive instrumentation, an 8-min baseline separation of the positional isomers (β - and γ -tocopherols) was easily achieved.

ACKNOWLEDGEMENTS

The authors wish to thank Hoffmann-La Roche, and Eisai for the generous gifts of tocopherols. We also wish to thank Mr. Phil Tao of V.P.I. & S. U. and Dr. David Colby of R. J. Reynolds Tobacco (Winston-Salem, N.C., U.S.A.) for their technical assistance and advice.

REFERENCES

- 1 R. G. Martinek, *Clin. Chem.*, 10 (1964) 1078.
- 2 A. Emmerie and E. Engel, *Rec. Trav. Chim. Pays-Bas*, 57 (1938) 1351.
- 3 H. G. Lovelady, *J. Chromatogr.*, 85 (1973) 81.
- 4 C. C. Tangney and J. A. Driskell, *Contraception*, 17 (1978) 499.
- 5 J. G. Bieri and E. L. Prival, *Proc. Soc. Exp. Biol. Med.*, 120 (1965) 554.
- 6 D. Carr, *Application Tip No. 87*, Varian Assoc., Palo Alto, Calif., 1971.
- 7 J. A. Schmit, R. A. Henry, R. C. Williams and J. F. Diekamn, *J. Chromatogr. Sci.*, 9 (1971) 645.
- 8 K. Abe, Y. Yuguchi and G. Katsui, *J. Nutr. Sci. Vitaminol.*, 21 (1975) 183.
- 9 P. J. van Niekerk, *Anal. Biochem.*, 52 (1973) 533.